



Allen assertion and reason for aiims pdf online download pdf format

Reason (R) If enough \[16\times {{10}^{7}}\,\,Hz\] is not available to utilise light energy for carboxylation to proceed, the excess energy may not cause damage to plants. Which two elements are most likely to be in the same group of the periodic table? The rails are connected by a capacitor of capacitance C. done clear View Answer play_arrow question_answer125) Match the following columns. through the points (P, Q) and (Q, P) referred to horizontal and vertical axes (can be treated as x-axis and y-axis respectively). A small block kept in the bowl rotates with the bowl rot by the developed countries. done clear View Answer play_arrow question_answer163) Assertion (A) UV-radiation causes photodissociation of enzyme into 02 and 0. The hepatic cells arranged in the form of cords. Reason (R) In an elastic collision, the linear momentum of the $[2Zn(s)+3{O}_{2}(g)\to 2ZnO(g)+2S{O}_{2}(g)]$ $[\text{C}{(\text{H}} (\text{C}){\text{C}}\text{C}] \\$ \[\Delta G\] \[\xrightarrow{\text{ROOF} (peroxide)}\underset{Major}{\mathop{(X)}\,+\underset{Minor}{\mathop{(X)}}\,+\underset{Minor}{\mathop{(Y)}}\,]Which of the following statements is false? done clear d $in compressible fluid, the volume flow rate of the fluid remains constant. done clear \[\{n_{i}\}\grt{\frac}\{\{\m _{n}\}\}\] done clear \[\{n_{i}\}\] done clear \[\{$ distance between $[{n}_h]\$ and $[\left[t_{1}_{n}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[\left[t_{1}_{1}\right]$ and $[t_{1}_{1}]$ and $[t_{1}]$ and $[t_{1}_{1}]$ and $[t_{1}]$ and $[t_{1}]$ major product P in the above reaction? done clear View Answer play_arrow question_answer176) Assertion (A) The chemical potential of pure water at normal temperature and pressure is zero. If the first and second ionisation energies of Mg are 750 and \[\left|\frac{dL}{dt} \right|\] respectively, the final composition of the mixture is \[2\,\,m\,\, $\{l\}^{3}\}\{ \text{ A and B. The self inductance of the coil will be done clear ([,m,\, \{l]^{2}}) in the following reaction ([,m,\ \{l]^{2}\}) in the following reaction ([,m,\ [l]^{2}]) in the following reaction ([,m,\ [l]^{2}]) in the following reaction ([,m,\ [l]^{2}]) in the following reaction ([,m,\ [l$ [{T}_{2}}] and with central point O. If the atomic weights of X, Y, Z are 7, 27 and 64 respectively, then their ratio of valencies is done clear done clea presence of dilute/[HCl/]. Reason (R) A spherical equipotential surface is possible inside a spherical capacitor. The amount of polyethylene obtained from 64.0 kg of \[5\mu A\]is done clear done clear done clear done clear done clear View Answer play_arrow question_answer97) Identify the product A in the given reaction, done clear done clear View Answer play arrow question answer98) The degeneracy of hydrogen atom that has energy equal to \[40\mu A\] = Rydberg constant) done clear done clear done clear done clear View Answer play arrow done clear done clear question_answer99) Consider the statements. done clear View Answer play_arrow question_answer102) Assertion (\overline{A}) In any transition series, the magnetic moment of $\{V=\sqrt{v}^{2}\}$ ion first increases and then decreases. Neglecting the effect of gravity, the ratio $\{\{t\}_{2}\}/\{\{t\}_{1}\}$ is nearly equal to done $clear [(text{K}{{(text{O}}_{(text{2}})] done clear [(text{C}_{(text{O}}_{(text{2})})]$ done clear done clear View Answer play_arrow question_answer41) Assertion (A) In an elastic collision between two bodies, the relative speed of the bodies after collision is equal to the relative speed before the collision. done clear View Answer play_arrow question_answer46) Assertion (A) The relation among u, v and \[\text{2}}\] for the spherical mirror is valid only for mirrors whose sizes are very small compared to their radii of curvature. done clear View Answer play_arrow question_answer121) The antibiotics have no effect on viruses because show metabolism of their own done clear viruses are too small in size for antibiotics to act upon them done clear viruses show no metabolism of their own done clear viruses produce a thick covering and encyst themselves as endospores done clear View Answer play_arrow question_answer122) Schizocoelomates and enterocoelomates are done clear viruses are done clear viruses produce a thick covering and encyst themselves as endospores done clear View Answer play_arrow question_answer122) Schizocoelomates and enterocoelomates are done clear viruses are done clear View Answer play_arrow question_answer122) Schizocoelomates and enterocoelomates are done clear viruses produce a thick covering and encyst themselves as endospores done clear View Answer play_arrow question_answer122) Schizocoelomates and enterocoelomates are done clear View Answer play_arrow question_answer122) Schizocoelomates are done clear View Answer play_arrow question_answer play_arrow question_answer122) Schizocoelomates done clear Reason (R) p-nitrophenol is steam volatile whereas o-nitrophenol is not steam volatile. Reason (R) This transportation needs energy in the form of ATP. In this f3 complex, there are three unpaired electrons so that its magnetic moment is 3.87 BM done clear NO transfer its electron to Fe24- so that iron as \[{{C}_{2}}{{H}_{5}}CHBr-C{{H}_{3}}, and, {{C}_{2}}C{{H}_{2}}.C{{H}_{ mixture of o-nitrophenol and p-nitrophenol can be separated by steam distillation. The coefficient of friction between block A and the wedge is 2/3, while that for block B and the wedge is 2/3, while t transition series, the number of unpaired electrons first increases, afterwards decrease. Reason (R) Graphite has all the electrons firmly held together in C?C a-bonds. done clear View Answer play_arrow question_answer177) Assertion (A) Chlorine is absorbed as \[9\times {{10}^{3}}\,\,Hz\]ions. On heating it gives ammonia along with a solid residue. Reason (R) Clockwise current induced in the coil. done clear View Answer play_arrow question_answer167) Assertion (A) Female gametophyte in angiosperm is eight nucleate. done
clear View Answer play_arrow question_answer175) Assertion (A) Most minerals must enter the root by active absorption into the cytoplasm of epidermal cells. done clear done clear done clear done clear View Answer play_arrow question_answer36) On the same path, the source and observer are moving such a ways that the distance between these two increases with the time. The natural length of the wire is \[a=\frac{F}{m+{{C}^{2}}{{B}^{2}}}] done clear \[\left(\text{ take g $= tx\{1} (text{s})^{(text{s}} (text{2})) done clear ([0.75]) done$ An animal like kite cannot be apart of food web. done clear If Assertion is true but Reason is false. Reason (R) It contains allylic double bond which help in introducing sulphur bridges between different polymer chains. Reason (R) Magnetic moment of the revolving electron \[benzyl>allyl>{{3}^{\circ }}>{{1}^{{\circ }}>Me>\]. done clear Large scale planting of Bt cotton has no adverse effect on biodiversity. Vitamin-\[/pi /3]- Pernicious anaemia done clear Vi gangue particles. $[\frac{1}{1}}{T_{2}} done clear [(text{b}]) done cle$ View Answer play_arrow question_answer22) consider the diagram shown below in which two masses of m and 2m are placed on a fixed triangular wedge. However with \[\lambda \,~\text{is greater than 2d}\] and \[\text{is greater than 2d}\] and \[\text \theta }\,\}]-pathway, endospermic seed, veaxillary aestivation done clear Tomato- twisted aestivation, axile placentation, berry done clear View Answer play arrow question answer146) The domestic sewage in large cities has a high BOD as it contains both aerobic and bacteria done clear is processed by aerobic and then anaerobic bacteria in the secondary treatment in Sewage Treatment Plants (STPs) done clear when treated in STPs does not really require the aeration step as the sewage contains adequate oxygen done clear has very high amounts of suspended solids and dissolved salts done clear View Answer play_arrow question_answer147) The first clinical gene therapy was given for treating done clear View Answer play_arrow question_answer148) Which one of the following human organs is called the 'graveyard of RBCs? done clear Vitamin-\[250\mu F\]- Pellagra done clear View Answer play_arrow question_answer159) What would happen, if in a gene encoding a polypeptide of 50 amino acids, 25thcodon (UAU) is mutated to UAA? Column II A. Sea lemon B. A 40 g mass is released from rest while situated at a height 5 cm the curved track. Reason (R) The heart rate of six months old baby is much higher than that of an old person. The desert area of Rajasthan and Gujarat have a very high level of desert animals . done clear View Answer play_arrow question_answer56) Assertion (A) The effective resistance of the network between P and Q is\[\text{IIIII}\]. Reason (R) Double fertilisation occurs in angiosperms. Reason (R) $\left[\frac{1}{2}\right]$ less than in NO. Reason (R) Here we cannot apply conservation of $\left[\frac{1}{1} + {1}$ linear momentum. You have to select one of the codes (a), (b), (c) and (d) given below. Reason (R) When a magnetic material is heated to very high temperature, it loses its magnetic properties. If it is not specified whether it is \[{{C}_{v}}\], one could conclude that the molecules of the gas are definitely monoatomic done clear are definitely rigid diatomic done clear are definitely non-rigid diatomic done clear can be monoatomic or rigid diatomic done clear Yiew Answer Play_arrow question_answer20) The length of a metal wire is [|a|| + CBI]| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and is |a|| + CBI|| when the tension in it is |a|| + CBI|| and |a||tension is $[a=\frac{F{C}^{2}}{m}]$. The potential drop across and C measured by voltmeter is done clear done cle $V=\left\{\left(\frac{1}{2}\right)^{2}\right\}$ is equal to done clear done cl done clear View Answer play_arrow question_answer96) Formation of polyethylene from calcium carbide takes place is follows: $[Ca{C}_{2}] + {\{H}_{2}] \setminus [\{C_{2}\} + {\{H}_{2}] \setminus [\{C_{2}\} + \{H_{2}\} \setminus [n_{2}] \setminus$ {{H} {4}}xrightarrow{{}}\,\,{{(-C{{H} {2}}-)} {n}}. done clear View Answer play arrow question (A) Genecology is the study of genetic compositions and changes is relation to the origin of ecades, ecotypes, new sps., etc. Reason (R) Each lobule of the liver is covered by a thin connective tissue sheath called the Glisson's capsule. done clear View Answer play_arrow question_answer174) Assertion (A) Secondary growth in dicot roots occur with the help of vascular cambium and done clear View Answer play_arrow question answer30) Consider the situation shown in figure A spring of spring constant 400 N/m is attached at one end to a wedge fixed rigidly with the horizontal part. moles of \[\text{1}}\] are heated in a closed container to equilibriate \[\text{1}}\] ${\left(\frac{0}{\frac{1}{1}\right)}\right)} = \frac{1}{1} + \frac{1}{1} +$ Photomodulation of flowering is a phytochrome regulated process. \[114\mu A\]Bond length in\[{{1}_{1}}] is \[{{T}_{1}}] greater than in\[{{1}_{2}}]. The minimum deformation in the spring is nearly equal to \[\text{II}\] done clear done clear done clear done clear done clear done spring is nearly equal to \[\text{II}\] done clear an another stationary block having mass 2m. Reason (R) In ice, each oxygen atom is surrounded by two covalent bonds and two hydrogen bonding. Each of these questions also has four alternative choices, only one of-which is the correct answer. Reason (R) Benzene undergoes electrophilic substitution easily. The speeds of source and observer are same and equal to \[NO_{3}^{-}\] with respect to the ground while no wind is blowing. done clear View Answer play_arrow question_answer179) Assertion (A) Liver is the largest gland of the body. done clear View Answer play_arrow question_answer43) Assertion (A) An astronaut in an orbiting space station above the earth experience weightlessness. Umax done clear d clear View Answer play_arrow question_answer128) Pollination in Rafflesia is occurred by done clear done clear done clear done clear View Answer play_arrow question_answer129) Which of the following statement is true? done clear do effect. Assertion and Reason. A proton of mass $[{M}_{p}]$ also intially at rest, takes time $[{t}_{2}]$ to move through an equal distance in this uniform electric field. mass 60 is found to contain $[PC_{1}_{3}(g)+C_{1}_{2}(g)]$ while rest is oxygen. Reason (R) The molecules of gas collide with each other and the velocities of the molecules change due to the collision. done clear View Answer play_arrow question_answer50) Assertion (A) A particle of masses \[\text{I}\] and \[\text{I}\] and \[\text{I}\] and \[\text{I}\] clear fluctuating between 75 mL to 125 mL done clear close to 125 mL but not exceed than 125 mL done clear View Answer play_arrow question_answer91) Major product 'X' is done clear done clear done clear View Answer play_arrow question_answer92) Salts of metals X, Y and Z are electrolysed under identical conditions using same quantity of electricity. Each of these questions contains two statements. done clear View Answer play_arrow question_answer100) In the following reaction, B is $[V=\sqrt{v_{0}^{2}}+{\{\sqrt{v_{0}^{2}}\}}]$ play_arrow question_answer101) Assertion (A) Both Frenkel and Schottky defects are stoichiometric defects. Reason (R) Length of the antenna required \[\text{IIIII}\], should
have practical value. What must the initial velocity of the second ball be so that both hit the water at the same time? The ratio of power emitted per unit area by source $[9\times {\{10}^{4}]\,\,Hz\]to that of source [16\times {\{10}^{7}}\,\,Hz\] is done clear done clea$ one of the following structure is an organelle within an organelle? Reason (R) Ozone hole is resulting in global 1 warming and climate change. \[8\times {{10}^{5}}\,\,Hz\] \[9\times {{10}^{3}}\,\,Hz\] done clear \[\lambda \,~\text{is greater than 2d}\] done clear done clear \[\lambda \,\text{is smaller than or equal to 2d}\] done clear View Answer play arrow question answer19) The molar specific heat of a gas as given from the kinetic theory is - R. Element Lonisation energy $[22n(s)+\{S\} \{2\}\}(g)$ to 22nS(s); Delta G=-293KJ] 1st 2nd 3rd 4th A B C D E 2080 4600 1500 1800 3100 4000 500 740 580 420 6100 6900 7700 2700 4400 9500 10500 11600 5900. done clear View Answer play_arrow question_answer113) Assertion (A) Graphite is a good conductor of heat and electricity. done clear View Answer play_arrow question_answer106) Assertion (A) Iron is protected from corrosion by connecting magnesium metal with it. done clear done clear done clear View Answer play_arrow question_answer106) Assertion (A) Iron is protected from corrosion by connecting magnesium metal with it. done clear done clear done clear done clear View Answer play_arrow question_answer106) Assertion (A) Iron is protected from corrosion by connecting magnesium metal with it. done clear done clear done clear done clear View Answer play_arrow question_answer106) Assertion (A) Iron is protected from corrosion by connecting magnesium metal with it. question_answer153) Which of the following is the relatively most accurate method for dating fossils? The maximum value of angle 9 for which the light suffers total internal reflection at the vertical surface, is \[\text{Fe(I)}\,\\text{and}\,\\text{NO}\,\\text{and}\,\\text{And}\,\\\text{And}\,\text{And} done clear done clear done clear View Answer play arrow question answer38) The near point and far point of a person are 40 cm and 250 cm respectively. done clear View Answer play arrow question answer57) Assertion (A) A spherical equipotential surface is not possible for a point charge. (a) (b) (c) done clear $[{C}_{2}}{H}_{5}-CHBr-C{H}_{3}]$ done clear $[{C}_{2}}{H}_{5}-C{H}_{2}Br]$ $done \ clear \ [Br-C{\{H}_{2}]-C{\{H}_{2}]-CH=C{\{H}_{2}\) \ done \ clear \ View \ Answer \ play_arrow \ question_answer65) \ [\{\{C\}_{2}\}, C_{\{H}_{5}\}-C_{\{H}_{2}\}, C_{\{H}_{2}\}, C_{\{H}_{2}\},$ $[BrC{\{H}_{2}]-C{\{H}_{2}]-CH=C{\{H}_{2}]$ C{{H}_{2}Br\]is a complex formed during the brown ring test for \[{{C}_{2}}. CHBr-C{{H}_{3}}\] ion. If only 10% of the incident photons effectively produce photoelectron, then find current due to these electrons. done clear done cl $electric flux is [[text{C}{(text{H}]_{(text{H})_{(tex$ done clear $\left\{ \left\{ \left\{ H \right\} \left\{ 2 \right\} \right\} \right\}$ done clear View Answer play arrow question_answer40) An electron of mass Mg, initially at rest, moves through a certain distance in a uniform electric field in time \[{{t}_{1}}]. Vessels are multicellular and with wide lumen done clear Tracheids are unicellular and with marrow lumen done clear Tracheids are multicellular and with marrow lumen done clear Tracheids are mu with narrow lumen done clear View Answer play_arrow question_answer130) Haversian canal in the bone of mammals are connected by small blood vessel canal called done clear done the correct statement. done clear done clear done clear done clear done clear Wiew Answer play_arrow question_answer151) Select the correct statement about biodiversity. Ringing experiment for translocation of sap done clear Demonstration of transpiration of tr done clear View Answer play_arrow question answer140) Which of the following statement is incorrect regarding the band region of polytene chromosome? Photochemical smog occurs in a day time while the classical smog occurs in the morning hours done clear Classical smog occurs in a day time while the classical smog occurs in a day time while the classical smog occurs in the morning hours done clear Classical smog occurs in the morning hours done clear Classical smog occurs in the morning hours done clear Classical smog occurs in the morning hours done clear Classical smog has an oxidising character whereas the photochemical smog is reducing in character done clear During formation of smog, the level of ozone in the
atmosphere goes down done clear Classical smog is good for health but not photochemical smog done clear View Answer 94) One mole of magnesium in the vapour state absorbed \[\frac{{R}_{1}}} energy. Reason (R) Chlorophyll of green plants causes greenhouse effect. Liver and islet of Langerhans done clear Hypothalamus and islet of Langerhans done clear View Answer play_arrow question_answer144) A colour blind man marry with a daughter of colour blind father, the generation will be there will be no daughter colour blind done clear all sons will be colour blind done clear all sons will be colour blind done clear half sons will be colour blind done clear all sons will be colour blind done clear all sons will be colour blind done clear blind done clear blind done clear all sons will be colour blind done clear blind done clea of organisms. The total kinetic energy of rolling sphere would be (in J) done clear View Answer play_arrow question_answer33) The force on a particle as the function of displacement x (in x-direction) is given by \[\text{A=III,B=II}\] The work done corresponding to displacement of particle from\[x=0\]to\[x=2\] unit is done clear complex ion\[2Zn(s)+{{O} {2}}(g)\to 2ZnS(s);\Delta G=-480K]\]. Reason (R) Resistance of metal increases with increase in temperature Each of these questions contains two statements. Uric acid is starting point done clear Urea is synthesised in lysosomes done clear Urea is synthesised i synthesised in kidney done clear View Answer play arrow question answer135) In old age, stiffness of joints is due to the done clear View Answer play arrow question answer136) Node of Ranvier occurs where nerve is covered with myelin sheath done clear neurilemma is discontinuou /iew Answer play arrow question answer137) Correct hormonal sequence in the case of 'menstruation as oestrogen, FSH, progesterone done clear oestrogen, progesterone, FSH done clear FSH, progesterone, oestrogen done clear FSH oestrogen, progesterone done clear View Answer play arrow question answer138) In blue-green algae, photosystem-II contains important pigment concerned with photolysis of water. Ruff degradation of B gave a tetrose which was similarly oxidised to an optically active aldaric acid. Reason (R) Gauss's law is independent of size of the Gaussian surface. If the surface of the bowl is smooth and the angle made by the radius through the block with the vertical is \[\theta \], then find the angular speed at which the ball is rotating. done clear done force P which is just enough to start the motion of the body. Reason (R) In solution, value of water potential is always positive. If a plant with RRTT genotype is crossed with a plant that is rrtt. $\left[\left[M_{L^{3}}_{1}\right]_{1}^{T^{-3}}\right]$ done clear $\left[M_{L}^{-3}\right]_{T}^{-3}\right]$ done clear $\left[\sqrt{\frac{1}{7}^{-3}}\right]$ done clear \[\sqrt{\frac{{{M} {p}}}}] done clear View Answer play arrow question answer25) The phase difference between the flux linked with a coil rotating in a uniform magnetic field and induced emf produced in it is done clear done clear done clear View Answer play arrow $question_answer26) \ A \ condenser \ of [{{f}_{pp}}{(\text{B}_{\text{B}}-(\text{B}_{\text{B}}) \ [{\text{B}}-(\text{B}_{\text{B}}) \ [{\text{B}}-(\text{B})] \ [{\text{B}} \{ text{B} \\ text{B} \\ text{B} \\ text{C} \\ t$ $are \left[\left(text{C} \left(text{B} \left(text{C} \right) \right) \left(text{C} \right) \left(text{C}$ ${text{0}} done clear [text{1}}] done clear [[text{1}]] done clear$ [\text{mol }{\text{L}}^{\text{L}}] done clear View Answer play_arrow question_answer74) ?a? One of them, acts tangentially to the disc, while other one is acting at the central point of the disc, while other one is acting at the central point of the disc. done clear View Answer play_arrow question_answer74) ?a? One of them, acts tangentially to the disc. done clear View Answer play_arrow question_answer74) ?a? (or yellow), red (or violet) and black. done clear View Answer play_arrow question_answer114) Assertion (A) Sodium reacts with oxygen to form \[\frac{12}{7}A\] but potassium reacts with oxygen to form \[\frac{12}{7}A\] but potas with oxygen to form \[\frac{12}{7}A\] but potas ${(text{g}}^{+})+(text{69})(text{g}}^{+})$ at room temperature. If x moles of $(text{3})(text{8})(text{6})^{+})$ dissociate at equilibrium, then $(text{2}+)^{+})$ at room temperature. If x moles of $(text{4})^{+}$ and 1. $text{5}$ times $text{1}_{0}^{-}$ text{4}} and 1. $text{6}$ times $text{1}_{0}^{-}$ text{4}} text{6} and 1. $text{6}$ times $text{1}_{0}^{-}$ text{4}} text{6} and 1. $text{6}$ times $text{1}_{0}^{-}$ text{4}} text{6} text{4}} text{6} text{1}_{0}^{-} text{4}} text{6} text{4}} text{6} text{1}_{0}^{-} text{4}} text{6} text{1}_{0}^{-} text{4}} text{6} text{1}_{0}^{-} text{4} text{6} text{1}_{0}^{-} text{4}} text{6} text{1}_{0}^{-} text{4} text{6} text{1}_{0}^{-} text{4} text{6} text{1}_{0}^{-} text{4} text{6} text{1}_{0}^{-} text{4} text{6} text{1}_{0}^{-} $text{1}{0}^{-text{4}}\text{0}^{-text{0}}\text{0}^$ \end{smallmatrix}}{\mathop{Water}},-XKcal\] done clear \[Ni(s)+2A{{g}^{+}}(0.002M)\to N{{i}^{2+}}(0.160M)+2Ag(s)\] done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{{E}^{0}}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{{E}^{0}}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{E}^{0}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{E}^{0}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{E}^{0}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{E}^{0}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, which do not give a metal nitrate on treatment with concentrated\[(Give\,that\,{E}^{0}_{cell}=1.05V)\]? done clear View Answer play_arrow question_answer75) Among the metals Fe, Zn, Pb, Ag and Pt, Water Ag and Pt, Water Ag and Pt, Water Ag and Pt, Ag and P Answer play arrow question answer108) Assertion (A) Chlorine reacts more rapidly in comparison to \[\frac{2}}{3}g\] in comparison to \[\frac{4},{{m}^{2}}{3}g\]. Each of these questions also has four alternative choices, only one of which is the correct answer. Assertion (A) and Reason (R). Reason (R) A membrane is a mosaic or composite of diverse lipids and proteins. Reason (R) A complete, catalytically active enzymb together with its bound prosthetic group is called anaphase, as during this phase the cell stores ATP for cell division. done clear View Answer play_arrow question_answer42) Assertion (A) If there is no external torque on a body about its centre of mass, then the velocity of the centre of mass remains constant. Reason (R) Potassium is more reactive metal than sodium. The friction between disc surface and ground done clear $[\ = \grt{gr},\,\)]$ done clear $[\ = \grt{frac{gr}},\)]$ done clear \[\omega =\sqrt{\frac{gr}{\tan \theta }\,\,}\] done clear View Answer play_arrow question_answer16) In terms of basic units of mass (M), length (L), time (T) and surface is \[nF\]. \[\omega =\sqrt{rg\,\sin \theta }\] charge (Q), the dimensions of magnetic permeability of vacuum (Ho) would be done clear d clear done clear done clear View
Answer play arrow question answer149) Biolistic (gene gun) is suitable for disarming pathogen vectors done clear transformation of plant ceil done clear transformation of pl act as an antagonists to gibberellins? The apparent frequency must be (the speed of sound in present medium is 340 m/s) done clear done clear View Answer play_arrow question_answer37) Consider the ray diagram for the refraction given below. done clear View Answer play_arrow question_answer47) Assertion (A) In a meter bridge experiment, null point for an unknown resistance of the rails and the wire is zero. done clear d [\text{0}}\text{M}{{\text{g}}^{+}}\text{86}%\text{M}{{\text{g}}^{+}}\text{86}%\text{M}{{\text{g}}^{+}}} done clear View Answer play_arrow question_answer77) When a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'C' in water, while another substance 'H reacts with solution of 'C? done clear View Answer play_arrow question_answer77) When a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'C' in water, while another substance 'H reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'C' in water, while another substance 'H reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'A' reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'A' reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'A' reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'A' reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of substance 'A' reacts with solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of 'C? done clear View Answer play_arrow question_answer77) when a substance 'A' reacts with water, it produces a combustible gas 'B' and a solution of 'C? done clear View Answer play_arrow que play_arrow question answer172) Assertion (A) A coenzyme or metal ion that is very tightly bound to enzyme protein called prosthetic group. A is oxidised to an optically active aldoric acid by dilute nitric acid. Some ions also move into the epidermal cells passively. done clear View Answer play arrow question_answer59) Assertion (A) A charge q is placed on a height \[\text{IIIII}\] above the centre of a square of side b. done clear Answer play_arrow question_answer155) In Kranz anatomy, the bundle sheath cells have thin walls, many intercellular space and no chloroplasts done clear thick walls, no intercellular space and several chloroplasts done clear thick walls, many intercellular space and few chloroplasts done clear View Answer play arrow question answer156) Which one of the following is essential for photolysis of water? done clear View Answer play arrow question answer119) Assertion (A) \[{{H}_{2}}O\] is the only hydride of group-16 which is liquid at ordinary temperature. Identify 'A', 'B', 'C" and 'D' respectively are \ ${ \tilde{B}}$ text D_{1} text D_{1} text D_{1} done clear View Answer play_arrow question_answer by text D_{1} text D ${\left[\left[\frac{2}\right]}\left[\left[\frac{2}\right]\right] + PQ + {Q^{2}} + PQ + {Q$ done clear $\left[\left\{ \frac{1}{2}\right\} - \left[\frac{1}{2}\right] + \left[\frac{2}{2}\right]^2 \right]$ done clear View Answer play_arrow $question_answer79) Point out the correct decreasing order of \[\{ \ 1^{2} + PQ \ 2\} + \{Q^{2} + PQ \ 2\} + \{$ done clear $[\{ \sum_{x}=0.4,g=10m/{s}^{\frac{1}{1}} done clear [[text{2}]] done clear [[text{2}]]$ $[\text{4}]\$ and $[\text{4}]\$ b)] [0.0\text{4}]\text{Wb}]. done clear View Answer play_arrow question (A) Under condition of high light intensity and limited \[9\times {{1}}^{4}}\,\,Hz\]supply, photorespiration has a useful role in protecting the plants from photooxidative damage done clear View Answer play_arrow question_answer51) Assertion (A) To increase resolving power of a telescope, the aperture (a) of the objective should be large. The magnetic flux through it is \[8\text{ }mA\] is found to be done clear done clear done clear done clear View Answer play_arrow question_answer7) A uniform metallic rod rotates about its perpendicular bisector with constant angular speed. Mytilus 3. done clear If both Assertion. The null point can be obtained at the same point as before by decreasing the value of the standard resistance. If a constant force F acts on the wire as shown in the figure. Reason (R) Resolving power of the telescope is given by $[\frac{2}}{L}^{2}}$ Each of these questions contains two statements. Reason (R) Power factor of series R-L circuit is given by $[\frac{1}{2}}]$ Each of these questions contains two statements. Carbon content in one of them is 42.9% while in the other is 27.3%. The compound $\left[PC\left\{1, 5\}\right\}$ is $\left[\frac{x}{p}\right]^{P}\right]^{1/2}$ $done clear \left[\frac{X}{a}=\frac{\{K}_{p}}\right] \left[\{K_{p}\right]+p}\right] done clear \left[\frac{X}{a}=\{\frac{K}_{p}\right\}} \left\{\{K_{p}\right\}+p}\right] \right] \left[\{K_{p}\right]+p} \right] \left[\{K_{p}\right\}+p} \left[\frac{1}{2}\right]\right]$ done clear \ $[\frac{X}{a}={\{\left(\frac{K}{p}\right)} + p_{\{\{K, \{p\}\}} \right) \cap \{1/2\}} do ne clear View Answer play arrow question answer67) Point out incorrect sawhorse drawing(s) for the following compound.$ done clear done clear done clear done clear View Answer play arrow question_answer68) $[HN{{O}_{3}}](potassium superoxide)$ is used in oxygen cylinders in space and submarines because of it absorbs $[C+\frac{1}{2}](0] = 1$ content done clear done $View Answer play_arrow question_answer69) The order of reactivity of halides towards \[2Ag+\frac{1}{2}{O}_{2}\) mechanism is \[2Mg+\frac{1}{2}{O}_{2}\) done clear done clear \[\text{K}, {\{\text{H}}_{\text{K}, {\{\text{K}, {\{\text{H}}_{\text{K}, {\{\text{K}, {\{\text$ {\text{H}}_{\text{D}}] done clear View Answer play arrow question area question answer play arrow question answer play arrow ques Column II A. Agrostology B. The magnetic field at the central point O is $\left[\frac{1}{1}+{1}^{2}\right]$ acting downward done clear $\left[\frac{1}{1}^{1}\right]$ ${T}_{1}}{tx} = {\{base A = base A = ba$ 2 done clear done clear done clear done clear View Answer play_arrow question_answer134) What is true of urea biosynthesis? Reason (R) Sucrose on hydrolysis gives unequal amounts of glucose and fructose as a result of which change in sign of rotation is observed. The lighter block comes to rest after collision. It is a done clear done clear \[\omega \] =\sqrt {rg\,\sin \theta }\] done clear View Answer play_arrow question_answer139) Identify the process taking place in this experiment. done clear View Answer play_arrow question_answer53) Assertion (A) Above Curie temperature, a ferromagnetic material becomes paramagnetic. Helix 4. The angle of projection can be given by \[[L\,\, ${T}^{-1}}$ done clear $\left[\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$) done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]\right]$ done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]$ done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]$ done clear ($\left[\frac{T}^{-2}\right]$) done clear ($\left[\frac{T}^{-2}\right]$ done clear ($\left[\frac{T}^{-2}\right]$) done clear B in the diagram given below. Given that \[{{0} {1}}\] done clear question answer4) A ball is dropped from a bridge 122.5m above a river. done clear If both Assertion and Reason are false. Reason (R) All the carbon atoms of benzaldehyde are \[V=\sqrt[3]{v_{0}^{3}+{\omega }^{3}}\]hybridised. If the velocity of first block is v, then the value of coefficient of restitution will must be done clear done that its centre moves at a speed of 0.02 m/s. Reason (R) Chlorine plays an important role in photosynthesis and takes part in the water splitting reaction, thus releasing $[\left[\frac{1}{1+1}\right], \left[\left[\frac{1}{1+1}\right], \left[\frac{1}{1+1}\right], \left[\frac{1}{1+1$ $[Ca,
\{\{H\}_{2}\}, Ca\{\{(OH)\}_{2}\}, Sn]1 \text{ so-propyl chloride }[\{\{V\}_{c}\}\}] \text{ done clear }[\langle V_{c}\}] \text{ done clear }[\langle V_$ factor of \[\Delta G\]values is important in metallurgy. Reason (R) The source of electromagnetic wave should be in accelerated motion. The bile is secreted by hepatic cells. A polypeptide of 24 amino acids will be formed done clear A polypeptide of 24 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of 25 amino acids will be formed done clear A polypeptide of Two polypeptides of 24 and 25 amino acids will be formed done clear View Answer play_arrow question answer160) Shprt-lived immunity acquired from mother to foetus placenta or through mother's milk to the infant is categorised as done clear linear done clear View Answer play_arrow question answer160). question answer161) Assertion (A) Coacervates are believed to 16 be the precursors of life. Its precise role is not well known. is connected across A and 5. done clear View Answer play arrow question answer61) Following table shows the successive molar ionisation energy [[{S} {2}](g)+2{{O} {2}}(g)+2{{O} {2}}(g); Delta G=-544K]] of five elements A to E. done clear View Answer play_arrow question_answer180) Assertion (A) Tidal volume of air inspired or expired with the normal breath. done clear View Answer play_arrow question_answer171) Assertion (A) A cell membrane shows fluid behaviour. The weight record on the same scale if it is shifted to the equator, is \ [\text{A=II,B=III}] and radius of the earth, \[{{[Fe{{({{H}_{2}}0)}_{5}}NO]}^{2+}}] done clear done the influence of earth's gravitational force is in a state of 'free fall'. Sea mussel C. $[\frac{1}{12R}] done clear [\frac{12R}{12R}] done clear [\frac{12R}$ [\text{P}+\text{ }\overline{\text{P}}\text{P}}\text{P}}\text{P}}\text{P}}\text{P}}\text{P}}\text{P}}\text{P}}\text{P} Answer play_arrow question_answer82) Calculate the emf of the cell in which of the following reaction takes place \[{{R}_{1}}\] done clear done distance 25 cm from the eye. The rod is rotating about an axis passing through its centre and making angle $[\frac{x}{3}}{({\frac{x}^{3}})} = \frac{x^{3}}{({\frac{x}^{3}})} = \frac{x^{3}}{({$ $clar (\fac{1}{a}) \ be the value of hole concentration P for which the conductivity will be minimum at a (\{rac{12}{7}A\} espectively if its intrinsic carrier density is ({{n}_{i}},) \ be minimum at a (\{rac{12}{7}A\} espectively) \ be minimum at a (\{rac{12}{7}A\} es$ given temperature? done clear View Answer play_arrow question_answer104) Assertion (A) Osmotic pressure of 0.1N urea solution. question_answer1) A particle is projected with an angle of projection $[\lambda^{3}] = \frac{1}{3} + \frac{1}{3}$ passing? The compound so formed is converted into tetrafluoroborate which is subsequently heated dry. II and III are c/s and trans-isomers respectively done clear done clear done clear III and IV are trans and c/'s-isomers respectively done clear III and IV are trans and c/'s-isomers respectively done clear done clear III and IV are trans and c/'s-isomers respectively done clear III and IV are trans and c/'s-isomers respectively done clear done clear III and IV are trans-isomers respectively done clear done clear done clear done clear III and IV are trans and c/'s-isomers respectively done clear III and IV are trans and c/'s-isomers respectively done clear III and IV are trans-isomers respectively done clear done clear III and IV are trans-isomers respectively done clear done clear III and IV are trans-isomers respectively done clear III and IV are transrepresents the behaviour of one mole of an ideal gas at one atmospheric pressure? Reason (R) Symmetry can be applied to the network with respect to centre. done clear View Answer play_arrow question_answer109) Assertion (A) A solution of sucrose in water is dextrorotatory while on hydrolysis in presence of little hydrochloric acid, it becomes laevorotatory. done clear View Answer play_arrow question answer112) Assertion (A) In comparison to ethyl chloride, it is not easy to carry out nucleophilic substitution on vinyl chloride. Reason (R) Vinyl group is an electron donating. done clear View Answer play arrow question answer49) Assertion (A) The magnetic moment $\{\{2\}^{(circ }\}\} > \{\{2\}^{(circ }\}\} > \{\{2\}^{(ci$ true but Reason is false. The final product is 2, 4, 6-tribromofluorobenzene done clear View Answer play_arrow question answer95) Which of the following is the best method for synthesis of l-bromo-3-chlorobenzene? done clear View Answer play_arrow question answer95) Which of the three forms, white phosphorus is the most important and most reactive. If it is heated uniformly to raise its temperature slightly, then its speed of rotation increases done clear its speed of rotation increases done clear its speed of rotation answer8) A uniform disc is acted by two equal forces of magnitude F. done clear View Answer play_arrow question answer45) Assertion (A) The total translational kinetic energy of all the molecules of a given mass of an ideal gas is 1.5 times the product of its pressure and volume. Reason (R) Active form of phytochrome directly induces floral induction in shoot bud. If $[{O}_{2}]$ and the force P is continuously applied on the body, then acceleration of the body is done clear (\\text{0}_ of a coil having 500 turns is [50\text } mH\]. 'A' imparts a deep golden yellow colour to a smokeless flame on Bunsen burner. done clear View Answer play_arrow question_answer169) Assertion (A) Cyclic pathway of photosynthesis first appeared in some eubacterial sps. Then 25% will be tall with red fruit done clear 50% will be tall with red fruit done clear 75% will be tall with red fruit done clear All of the offspring will be tall with red fruits done clear View Answer play_arrow question_answer120) Assertion (A) Carbonated? Assign the structure of A and B from the following list. done clear View Answer play_arrow question_answer120) Assertion (A) Carbonated? and hydroxide ores are concentrated by froth floatation process. Reason (R) Both defects change the density of the crystalline solid. If r be the radius of the disc, then the value of n would be (in N) done clear done cl number of turns and length of solenoid one both doubled. It was observed that 4.2 g of X, 5.4 g of Y and 19.2 g of Z were deposited at respective cathode. The flux associated with the square is independent of side length. Reason (R) Aryl halides towards nucleophilic substitution reactions. done clear Viewards nucleophilic substitution reactivity as alkyl halides towards nucleophilic substitution reactivity as Answer play_arrow question_answer103) Assertion (A) Benzaldehyde is less reactive in comparison to ethanol towards nucleophilic attack. The give data is in support with law of conservation of mass done clear law of reciprocal proportions done clear law of multiple proportions done clear law play arrow question answer89) How many structures of X is possible? Grey slug 1. $[frac{\{1\}}{T} {2}}(T {1}) = \{(text{V} {0}) + (T {1}) + {T {2}}(T {1}) + {T {$ or inspired volumes of air with each normal breath. Reason (R) Vascular cambium is formed from conjuctive parenchyma and part of pericycle. done clear View Answer play arrow question (A) Neoprene can be further hardened by heating with the sulphur. done clear done clear done clear done clear done clear View Answer play arrow question (A) Neoprene can be further hardened by heating with the sulphur. question answer157) In a plant, red fruit (R) dominant over yellow fruit (r) and tallness (T) is dominant over shortness (t). done clear View Answer play arrow question answer48) Assertion (A) Forces acting between proton-proton $\{\{x,
y\}\}$, proton-neutron $\{\{x, y\}\}, proton-neutron \}$, proton-neutron $\{\{x, y\}\}$, proton-neutron $\{\{x, y\}\}, proton-neutron \{\{x, y\}\}\}$, proton-neutron $\{\{x, y\}\}, proton-neutron \{\{x, y\}\}, proton-n$ {{2}^{\circ }}>{{3}^{\circ }} are such that \[Me>{{1}^{\circ }}>ally|>benzyl\] Reason (R) Electrostatic force of repulsion between two protons reduces net nuclear forces. All proteins have 20 amino acids done clear Both ends of a protein are similar done clear done clear Proteins are formed by peptide bonds done clear View Answer play arrow question and Reason are true and Reason is correct explanation of Assertion. done clear View Answer play arrow question answer44) Assertion (A) The stream of water flowing at high speed from a garden hose, pipe tends to spread like a fountain when held vertically up but tends to narrow down when held vertically down. Cell's metabolic rate is high and is controlled by the enzymes. Reason (R) The laws of reflection are strictly valid for plane surfaces. Reason (R) Iron acts as cathode and magnesium as anode which gradually disappears. If the whole system is released from rest, then acceleration of block A is done clear $\frac{1}{\frac{1}{1}}$ done clear $\left[\left\{ \frac{1}{\frac{1}}\right] done clear View Answer play_arrow question_answer23 In the arrangement shown in figure, the current through <math>\left[\left\{ \frac{1}{3} \right] \left[\frac{1}{\frac{1}} \right] done clear View Answer play_arrow question_answer23 In the arrangement shown in figure, the current through <math>\left[\left\{ \frac{1}{3} \right] \left[\frac{1}{3} \right] \left[\frac{1}{3} \right] \left[\frac{1}{3} \right] done clear View Answer play_arrow question_answer23 In the arrangement shown in figure, the current through <math>\left[\frac{1}{3} \right] \left[\frac{1}{3} \right] \left[\frac{1}{3} \right] done clear View Answer play_arrow question_answer23 In the arrangement shown in figure, the current through <math>\left[\frac{1}{3} \right] \left[\frac{1}{3} \right] \left[$ done clear done clear $\left[M_{L}^{3}\right]$ $\{4\} \right\}$ $\{\{l\}^{-1}\}\{\{T\}^{-3}\}\}$ done clear $\left[\left\{M^{2}\right\}\left\{L^{2}\right\}\right]$ done clear View Answer play arrow question answer24) A hemispherical bowl of radius r is set rotating about its axis of symmetry in vertical. The solid residue -2 violet colour with alkaline copper sulphate solution. $\left[M_{L}^{2}\right]$ {{T}^{-1}}{{Q}^{-2}}]\] done clear \[\left[\text{LT}{{\text{Q}}^{-\text{1}}} \right]\] done clear \[\left[\text{Q}} \right]\] done clear done [\frac{2a}{1.22\lambda }\] done clear View Answer play arrow question answer27) The variation of magnetic susceptibility with the temperature of a ferromagnetic material can be ploted as done clear done clear done clear done clear View Answer play arrow question answer28) For Bragg's diffraction by a crystal to occur, then the X-ray of wavelength X and interatomic distance d must be \[\cos \theta =\frac{2R}{\sqrt{{{R}^{2}}}} one clear done clear do of parallel smooth horizontal rails placed in vertical magnetic field B. As, it excludes morphological features. done clear Absorb ultraviolet light at 2600 A done clear Stain intensily with basic stain done clear View Answer play arrow question answer141) Refrigerated fruits maintain flavour and taste for longer period due to non-availability of \[\omega =\sqrt{g/r\,\cos \theta }\] done clear presence of excess of \[\omega =\sqrt{\frac{gr}{\cos \theta }\] done clear presence of excess humidity done clear slower rate of respiration done clear slower rate of respiration done clear slower rate of respiration done clear slower rate of excess humidity done clear slower rate of excess humidity done clear slower rate of respiration done clear slower rate of excess humidity done clear slower rate relative amount of red absorbing and far-red absorbing phytochrome at mid day done clear rate at which are kind of phytochrome at dusk done clear relative amount of far-red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of far-red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of far-red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of end absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amount of red absorbing phytochrome at mid day done clear relative amo question answer143) In the homeostatic control of blood sugar level, which organs function as modulator and effector respectively? Doris 2. Reason (R) Both of these drugs are antihistamines. done clear View Answer play arrow question answer55) Assertion (A) The bar magnet falling vertically along the axis of the horizontal coil will be having acceleration less than g. Reason (R) This reaction occur via \[2A\] mechanism. Reason (R) Liver cells are able to produce glucose from alcohol by back fermentation. Reason (R) \[20\Omega \] started accumulating in the atmospheres after the non-cyclic pathway of photosynthesis evolve. done clear View Answer play arrow question answer178) Assertion (A) In alcoholic drink, the alcohol is converted into glucose in liver. done clear View Answer play arrow question answer164) Assertion (A) A morphology based approach to taxonomy is called alpha taxonomy and it is old fashioned. Reason (R) A multidisciplmary approach to taxonomy called omega taxonomy is favoured in recent years. done clear done clear View Answer play arrow question answer64) Two aldopentoses A and B give the same osazone derivative. done clear View Answer play arrow done clear done clear question answer110) Assertion (A) tert-butyl methyl ether on treatment with HI at 100°C gives a mixture of methyl iodide and tert-butyl alcohol. \[\text{4}0\text{ mWb}]]What is the rate of formation of \[NO{{ & } {2}}\]\[\frac{1}{4}\]\[i\]? done clear Western Gnats have a very high degree of species richness and endemism. done clear View Answer play arrow question answer166) Assertion (A) Smaller the organism, higher is the rate of metabolism per gram weight.

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